

Exploring the Chemical Composition of Day vs Nighttime Particulate Matter

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Background

- Poor air quality was responsible for approximately 8 million premature deaths in 2019¹
- Fine particulate matter (less than 2.5 microns-PM 2.5) contributes to poor air quality-which in cities can lead to the formation of smog
- PM 2.5 particles are created through atmospheric chemical reactions involving gasses
- PM 2.5 particles are small enough to travel deeper into the lungs and be absorbed into the bloodstream
- The Northern Front Range of Colorado has over 30 days a year where the air quality index (AQI) is over 100- which is unhealthy. Nitrates and nitrated products can significantly contribute to the level and type of pollution in populated areas²

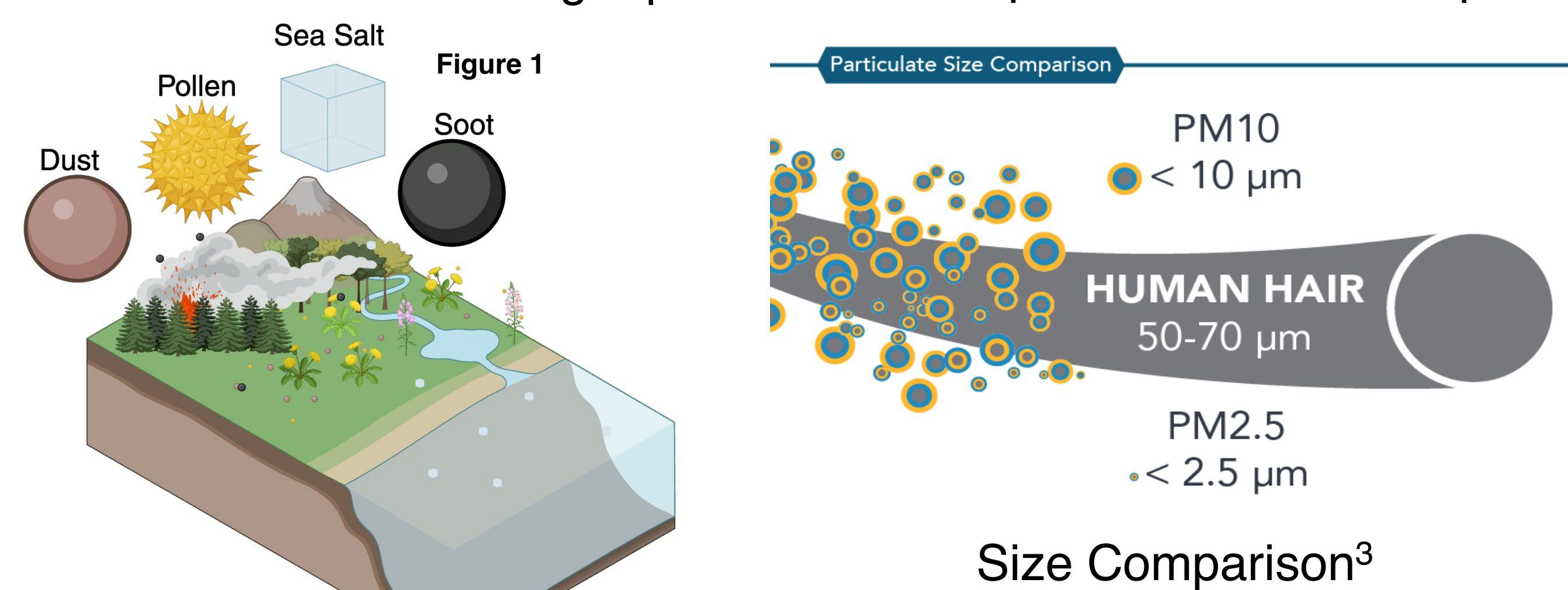
Introduction

- This study examined the impacts of air pollution from urban to rural environments in the Colorado Front Range
- We studied the chemical composition of PM 2.5 and used the chemical composition to explore the impacts of air pollution
- Nitrates are typically dominant at night; however, in urban areas, human emissions can cause nitrates to be present in the daytime
- We expected nitrates to be highest in Denver, lower in Boulder and lowest at the Mountain Research Station (MRS)

Particulate Matter Formation

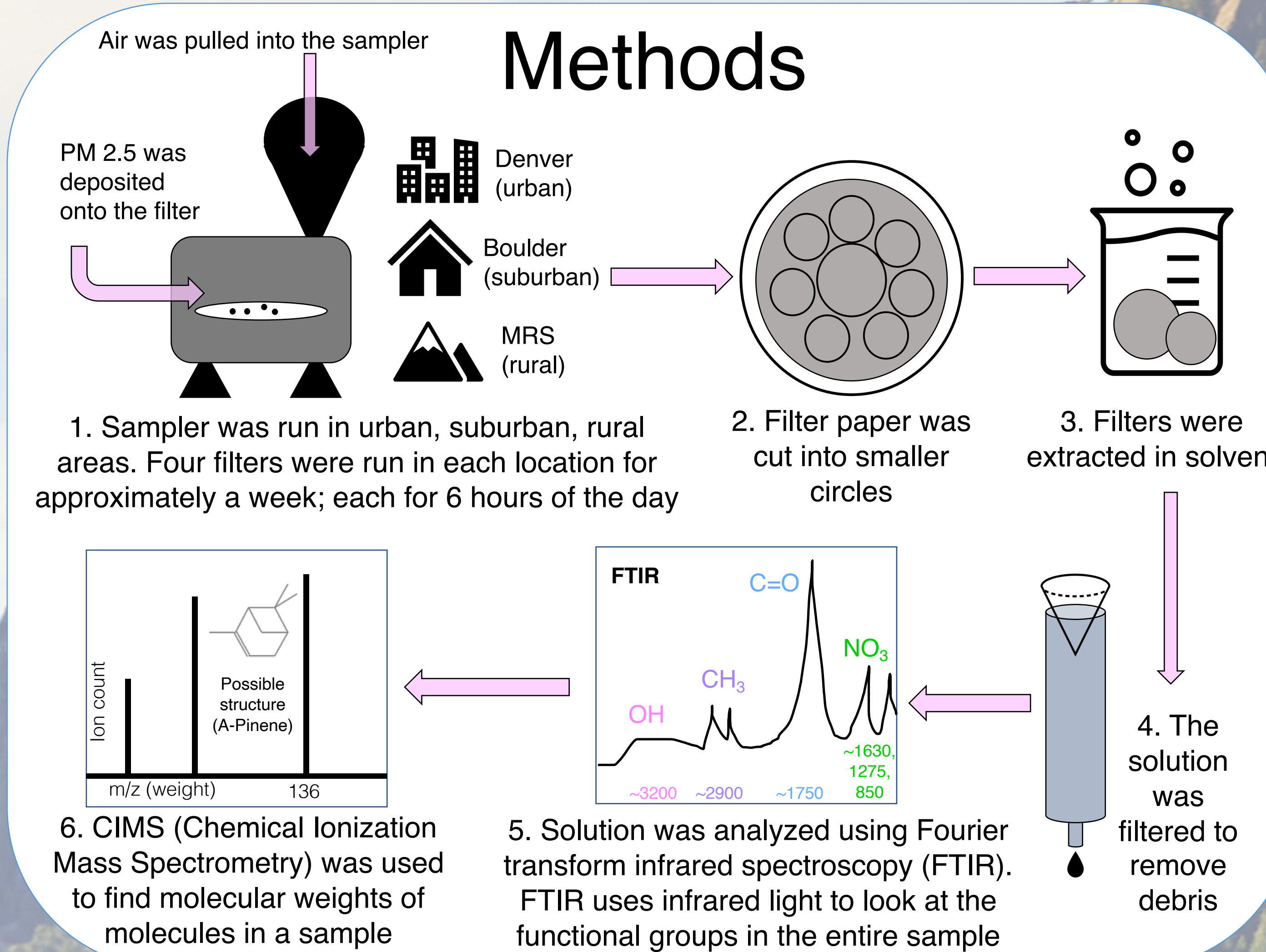
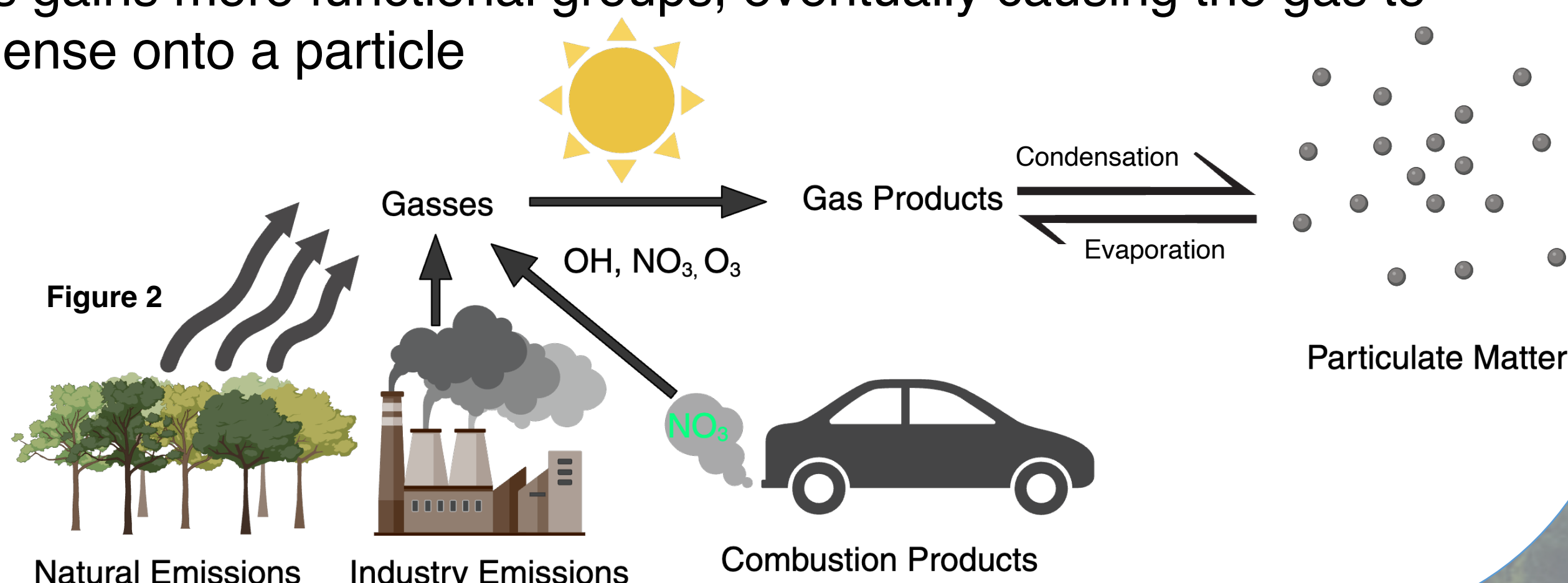
PM 10

PM 10 is formed when larger particles are suspended in the atmosphere



PM 2.5

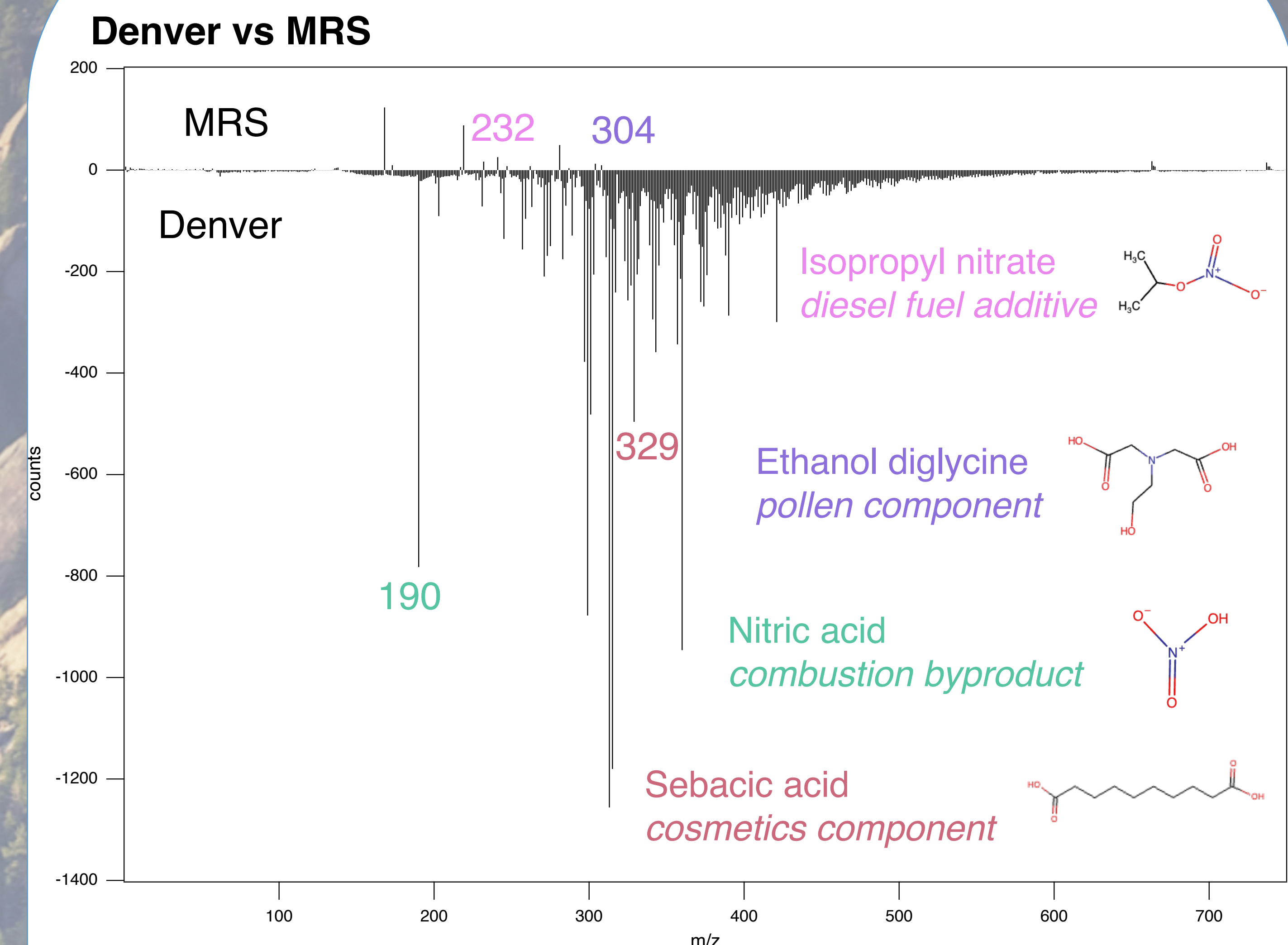
PM 2.5 is formed when gases react with each other. With each reaction a gas gains more functional groups, eventually causing the gas to condense onto a particle



Methods

1. Sampler was run in urban, suburban, rural areas. Four filters were run in each location for approximately a week; each for 6 hours of the day
2. Filter paper was cut into smaller circles
3. Filters were extracted in solvent
4. The solution was filtered to remove debris
5. Solution was analyzed using Fourier transform infrared spectroscopy (FTIR). FTIR uses infrared light to look at the functional groups in the entire sample
6. CIMS (Chemical Ionization Mass Spectrometry) was used to find molecular weights of molecules in a sample

CIMS Results



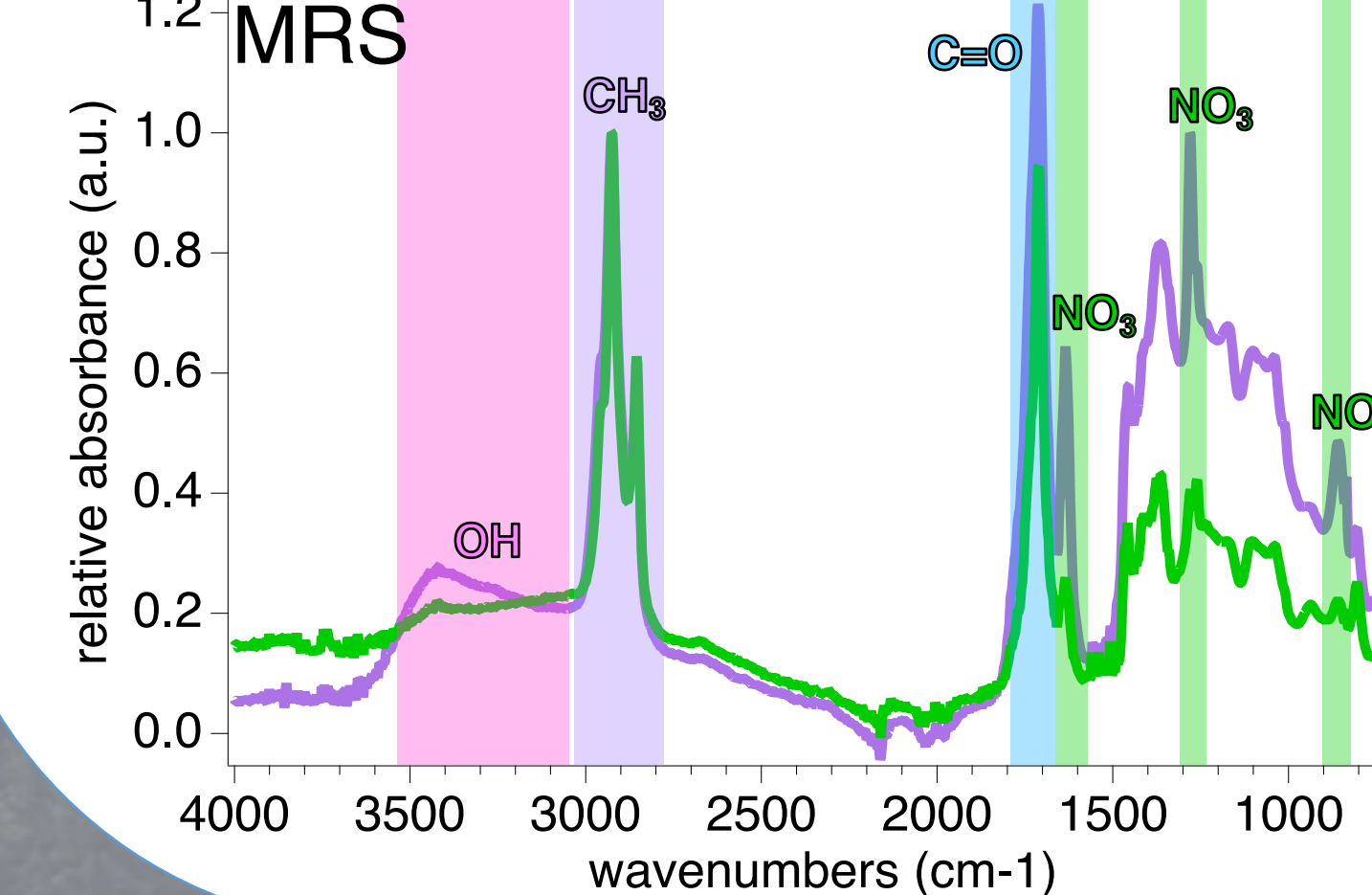
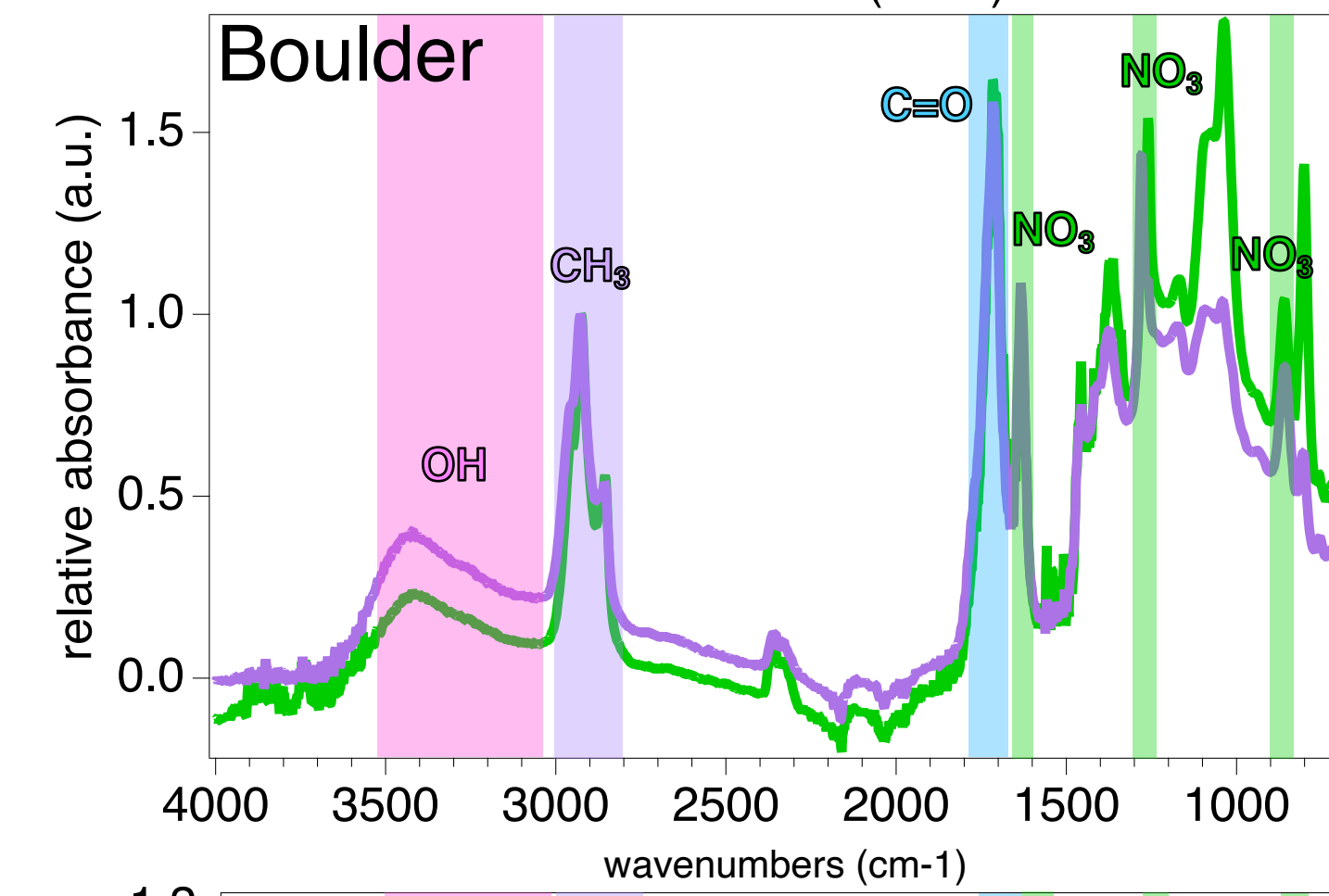
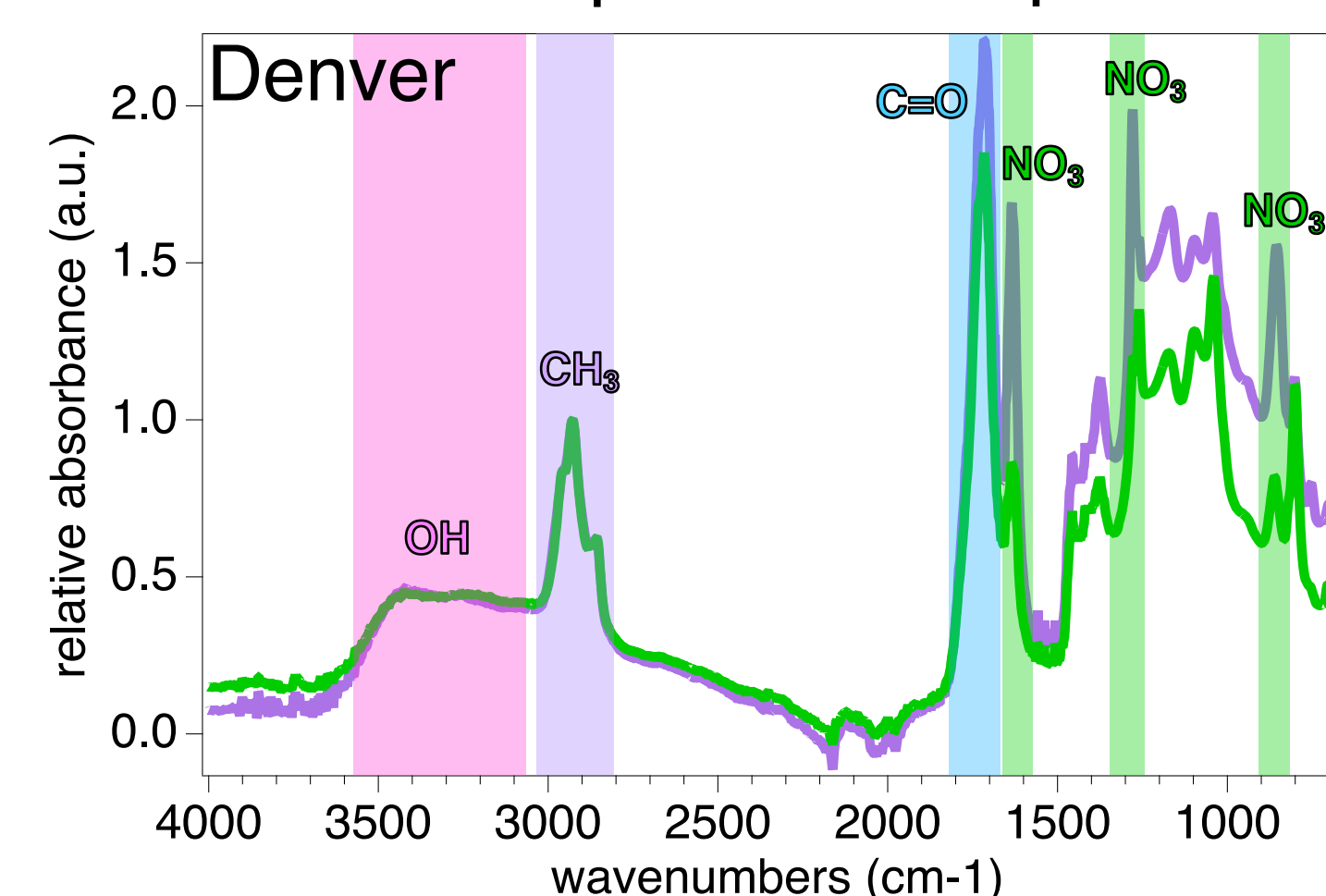
General Findings

- Isopropyl nitrate and ethanol diglycine were largely unique to the MRS
- Most products were less than 500 m/z which is consistent with semi-volatile molecules typically found in PM

FTIR Results

Daytime/Nighttime FTIR

Day: 10am-4pm (Green)
Night: 10pm-4am (Purple)



Nitrate/Carbonyl Ratio

	Day	Night	Change
Denver	0.465	0.765	160%
Boulder	0.524	0.687	130%
MRS	0.274	0.537	200%

- A nitrate (NO₃)/carbonyl (C=O) ratio helps determine relative abundance of nitrates. A higher number means more nitrates were present

General Findings

- Nitrates were more abundant at night
- Products were more functionalized in higher populated areas
- Higher functionality in Denver and Boulder indicates a higher concentration of reactive gasses and OH, NO₃, and O₃ relative to MRS

Conclusion

- Chemical functionality increased as the population increased
- More nitrated products were found in nighttime samples
- CIMS data indicates that the MRS had unique products that were not present in Denver
- Future work could potentially focus on differences in PM composition seasonally; and including quantitative analysis of functional groups to gain a better understanding of pollution composition
- Future work could also look into the impacts of wildfires on particulate matter formation; particularly in urban areas

References

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4. Figures 1 and 2 created with Biorender.com
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